

Gallium-Catalyzed Domino Arylation/Oxycyclization of Allenes with Phenols

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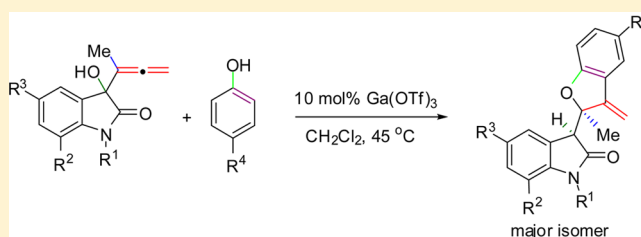
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Supporting Information

ABSTRACT: The synthesis of dihydrobenzofuran-appended oxindoles has been accomplished taking advantage of an unprecedented reaction between allenols and phenols under metal catalysis.

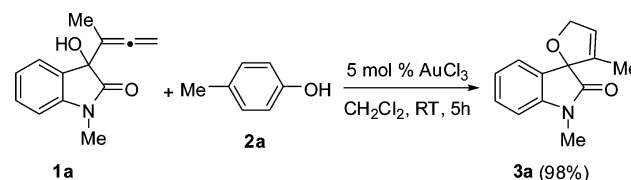


The dihydrobenzofuran motif is present in a wide variety of natural products and biologically relevant compounds,¹ and the synthesis of this heterocycle is of current interest.² Numerous reports are available on metal-catalyzed cyclization or cross-coupling reactions of functionalized allenols.³ In contrast, such reactions that involve the coupling of the allene moiety and a phenol are scarcely accessible in literature.⁴ Despite the fact that phenols are readily available chemicals, their use is problematic due to selectivity issues. Recently, we have successfully reported selective transformations of both indolinone-tethered allenols⁵ and phenols.⁶ We envisioned that a different behavior of the allenol moiety might be achieved utilizing a phenol as a coupling partner. Herein, we present a gallium-catalyzed coupling–cyclization between phenols and indolinone-tethered allenols toward the preparation of dihydrobenzofuran-linked oxindoles.

To explore the possibility of an allene–phenol coupling, allenol **1a** and phenol **2a** were initially chosen. The AuCl₃-catalyzed reaction of allenol **1a** and 4-methylphenol **2a** afforded the spirocyclic 2,5-dihydrofuran **3a** (Scheme 1). Hence, the hydroxy group in allenol **1a** exclusively suffers a *S-endo* oxycyclization reaction,⁷ without the participation of the phenol moiety.

When a π acid such as AuCl₃ is used, it might coordinate with one of the allene double bonds via a monodentate mode. When a Lewis acid is employed, it might coordinate with the OH of the allenol moiety via mono- or bidentate modes. This activation might generate a carbocation intermediate. Consequently, we decided to use a main-group salt instead of a transition-metal derivative. Happily, after assessing various metal catalysts, we found that a catalytic amount of metal triflate specifically promoted the domino allenol–phenol coupling reaction (Table 1). The domino reaction took place

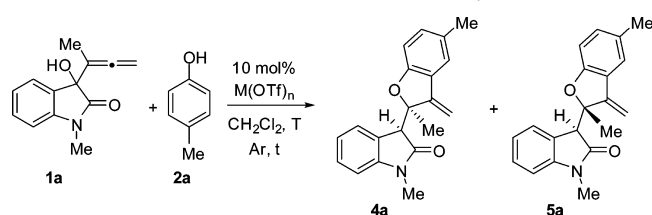
Scheme 1. Nonproductive Gold-Catalyzed Reaction of Indolinone-Tethered Allenol **1a** and Phenol **2a**



at 40 °C under M(OTf)_n catalysis, which specifically promoted the generation of the desired dihydrobenzofuran-appended oxindole scaffold, and the domino addition–cyclization reaction took place readily.⁸ Diastereoselectivities were modest, in all cases giving rise to mixtures of adducts **4a** and **5a**. Based on conversion and isolated yields, gallium(III) triflate proved to be the most efficient Lewis acid catalyst (Table 1, entry 12).^{9,10} An experiment using molecular sieves as an additive led to comparable results, thus indicating that this transformation could be efficiently catalyzed by metal triflates. A screening of solvents (acetonitrile, tetrahydrofuran, toluene) revealed that the reaction is best performed in dichloromethane. A Brønsted acid such as trifluoromethanesulfonic acid (TfOH) was also tested. The corresponding rearranged α,β -unsaturated ketone was the major reaction product under stoichiometric TfOH conditions. However, the use of 10 mol % TfOH afforded adducts **4a** and **5a** along with a minor rearranged ketone, but did not promote the reaction to completion, with some unreacted starting material remaining. Comparatively, the use of TfOH led to limited reactivity.

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Table 1. Reaction between Indolinone-Tethered Allenol **1a** and Phenol **2a** under Modified Metal-Catalyzed Conditions

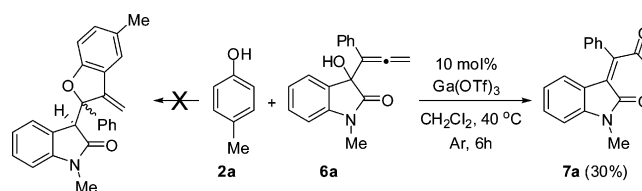
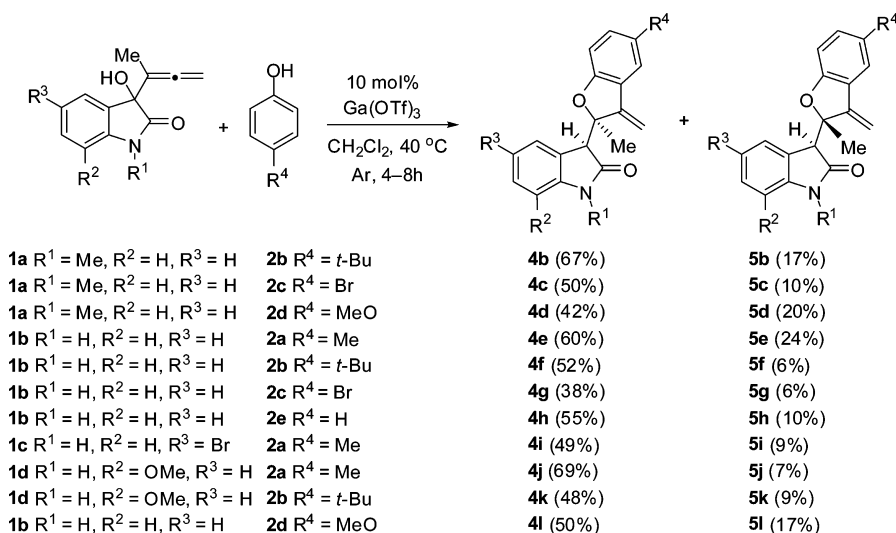
entry	catalyst	temperature (°C)	time (h)	4a:5a yield (%) ^a
1	In(OTf) ₃	40	11	25/8
2	Zn(OTf) ₂	40	12	22/7
3	Fe(OTf) ₃	40	10	29/9
4	Yb(OTf) ₃	40	12	20/6
5	Bi(OTf) ₃	40	1	26/19
6	Bi(OTf) ₃	30	3	23/17
7	Bi(OTf) ₃	130/sealed tube	0.5	6/4 ^b
8	Bi(OTf) ₃	80/microwave	0.5	19/14
9	Bi(OTf) ₃ /PTSA	40	1	11/15
10	Ga(OTf) ₃ ^c	40	30	23/13
11	Ga(OTf) ₃ ^d	40	12	30/17
12	Ga(OTf) ₃	40	4	35/20

^aYield of pure, isolated product with correct analytical and spectral data. ^bDecomposition of the starting allenol **1a** was observed in appreciable extension. ^cCatalyst loading of 2 mol %. ^dCatalyst loading of 5 mol %.

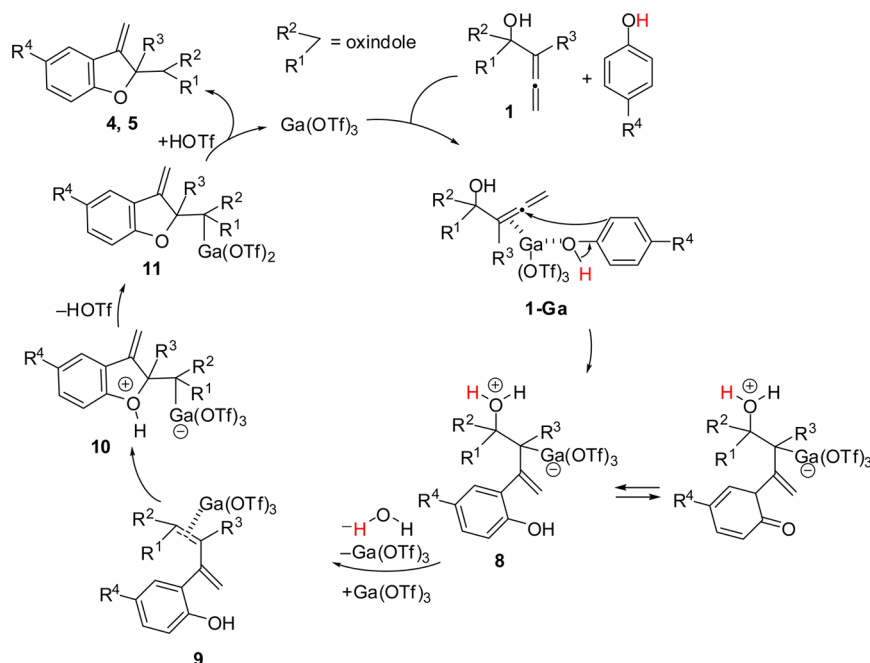
With the optimized reaction conditions in hand we then examined the scope and generality of the Ga-catalyzed method. Thus, various methyl-substituted allenols and phenols were reacted to give a range of dihydrobenzofuran-appended oxindoles **4** and **5** (Scheme 2).¹¹ The two products obtained from the coupling reaction are not identical but rather stand in an epimeric relationship at the allylic position. Although the diastereoselectivity of the reaction between **1a** and **2a** is poor under the conditions (Table 1), efforts for further improvements were not in vain. Interestingly, the use of both NH-free allenols **1b–d** and phenols bearing bulkier substituents (such as Br or *t*-Bu) resulted in improved diastereoselectivities (Scheme 2). The influence of the electronic nature of the substituents at

the phenol reactant was first examined by submitting 4-methoxyphenol **1d** to various allenols **1**. In the event, phenol **1e** reacted well and the corresponding adducts **4d**, **4l**, **5d**, and **5l** were obtained. Unfortunately, a negative result was observed using either 4-hydroxybenzotrile or methyl 4-hydroxybenzoate, which indicated that electron-withdrawing groups on the aromatic ring were not compatible with this catalytic arylation/oxycyclization reaction sequence. The *R_f* values of the two dihydrobenzofuran isomers **4a–l** and **5a–l** were very close to each other. Fortunately, the diastereomeric adducts **4** and **5** could be separated by flash chromatography, thus readily providing two structurally complex and valuable heterocyclic products. However, characterization of minor adducts **5c**, **5d**, **5f–i**, **5k**, and **5l** was performed on a mixture containing small amounts of their major counterparts **4**. Because most of the reactions were conducted in a 50 mg scale, it was desirable to scale-up the procedure. Worthy of note, when we performed a 4 mmol-scale reaction starting from allenol **1a** and phenol **2a**, adducts **4a** and **5a** were isolated in a combined yield of 59%, which is slightly higher than that achieved at a smaller scale during the scope study. The structure of dihydrobenzofuran **5a** was unambiguously confirmed with the help of an X-ray diffraction analysis on suitable crystals of this compound (Figure 1 in the Supporting Information).¹²

Next, the reaction of aryl-substituted allenol derivatives and phenols was examined. To test the reactivity of aryl allenols **6**, we started the initial investigation on the gallium-catalyzed reaction of phenyl-substituted allenol **6a** under otherwise identical reaction conditions used for its methyl-counterpart **1a**. In the event, it was found that substrate **6a** was exclusively transformed into the rearranged product **7a** (Scheme 3).¹³

Scheme 3. Reaction of Phenyl-Substituted Allenol **6a** under Metal Triflate Catalysis**Scheme 2.** Synthesis of Dihydrobenzofuran-Appended Oxindoles **4** and **5** through Domino Addition–Cyclization Reaction under Gallium Catalysis

Scheme 4. Mechanistic Explanation for the Ga(III)-Catalyzed Synthesis of Dihydrobenzofurans 4 and 5 from Allenols 1 and Phenols



Noticeably, despite the above-mentioned ability (Schemes 1 and 2) of metal triflate based catalysis for the coupling reaction of allenols **1** and phenols into dihydrobenzofurans **4/5**, no traces of oxacycles were detected using gallium(III) triflate as a promoter. Possibly, attempts to use aryl-substituted substrates **6** proved to be unsuccessful for the construction of the corresponding dihydrobenzofurans because of both unfavorable steric factors and a direct interaction of the π -aromatic system with the metal center from the catalyst.

A possible pathway for the gallium-catalyzed generation of dihydrobenzofuran-linked oxindoles **4** and **5** is outlined in Scheme 4. Initially, $\text{Ga}(\text{OTf})_3$ acts as a Lewis acid interacting with the allene and phenol moieties simultaneously via a bidentate mode. The formation of a complex **1-Ga** through both π -coordination of the metal to the allene group of allenols **1** and σ -coordination to the hydroxy group of phenols **2** may be involved. Subsequent nucleophilic addition of the phenol moiety at the sterically less hindered carbon center in complex **1-Ga** would lead to a ketone intermediate, which after tautomerization afforded the more stable enol-form **8**. Then, an elimination of water and $\text{Ga}(\text{OTf})_3$ from zwitterionic intermediate **8** occurs to generate gallium species **9** through η -coordination of the gallium salt to the more substituted double bond. Species **9** suffers an intramolecular chemo- and regioselective 5-*exo*-trig oxycyclization reaction to produce zwitterionic dihydrobenzofurans **10**. This nucleophilic attack from the *O*-phenol site occurs as a result of the stability of the intermediate oxonium type cation **10**. Loss of HOTf in zwitterionic dihydrobenzofurans **10** generates neutral metal species **11**. Finally, protonolysis of the carbon–gallium bond of **11** liberates dihydrobenzofurans **4** and **5** with concomitant regeneration of the Ga(III) catalytic species (Scheme 4).

In conclusion, the synthesis of dihydrobenzofuran-linked oxindoles has been accomplished taking advantage of an unprecedented reaction between allenols and phenols under Lewis acid catalysis. Probably, $\text{Ga}(\text{OTf})_3$ merges allene and phenol moieties simultaneously via a bidentate mode.

EXPERIMENTAL SECTION

General Methods. NMR spectra were recorded at 25 °C on 700 or 300 MHz spectrometers: ^1H NMR (300 or 700 MHz) and ^{13}C NMR (75 or 175 MHz). NMR spectra were recorded in CDCl_3 or C_6D_6 solutions, except if otherwise stated. Chemical shifts are given in ppm relative to TMS (^1H , 0.0 ppm), or CDCl_3 (^{13}C , 76.9 ppm), or C_6D_6 (^{13}C , 128.4 ppm). Low and high resolution mass spectra were acquired on a QTOF LC/MS spectrometer using the electronic impact (EI) or electrospray (ES) mode.

Typical Procedure for the $\text{Ga}(\text{OTf})_3$ -Catalyzed Coupling Reaction of α -Allenols **1 and Phenols **2**. General Procedure for the Preparation of Dihydrobenzofuran-Appended Oxindoles **4** and **5**.** To a solution of the appropriate allenol **1** (0.46 mmol) in dichloromethane (5 mL) at room temperature, $\text{Ga}(\text{OTf})_3$ (0.046 mmol) was added under an argon atmosphere and stirring was continued for 5 min. Then, the corresponding phenol **2** (1.28 mmol) was added and stirring was continued for another 5 min. Then, the reaction mixture was stirred at reflux temperature until the starting material disappeared as indicated by TLC. After filtration through a pad of Celite, the mixture was concentrated under vacuum and purified by flash column chromatography, eluting with ethyl acetate/hexanes or dichloromethane/ethyl acetate mixtures. Spectroscopic and analytical data for pure forms of compounds **4** and **5** follow.

Preparation of Dihydrobenzofuran-Appended Oxindoles **4a and **5a**.** From 32 mg (0.14 mmol) of α -allenol **1a**, and after chromatography of the residue using hexanes/ethyl acetate (5:1) as eluent, 15 mg (35%) of the less polar compound **4a** and 8 mg (20%) of the more polar compound **5a** were obtained.

Dihydrobenzofuran-Appended Oxindole **4a.** Yellow oil (15 mg, 35%); ^1H NMR (300 MHz, CDCl_3 , 25 °C) δ : 7.19 (d, $J = 7.5$ Hz, 1H), 7.12 (t, $J = 7.7$ Hz, 1H), 6.99 (m, 1H), 6.91 (s, 1H), 6.78 (d, $J = 8.3$ Hz, 1H), 6.72 (m, 2H), 5.09 and 4.76 (s, each 1H), 3.88 (s, 1H), 3.20 (s, 3H), 2.20 (s, 3H), 1.94 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3 , 25 °C) δ : 173.1, 158.4, 148.0, 144.2, 131.4, 129.7, 128.1, 126.3, 126.1, 125.1, 122.2, 121.0, 109.5, 107.5, 100.0, 91.0, 54.0, 26.1, 25.9, 20.7; IR (CH_2Cl_2 , cm^{-1}): ν 1710, 1611, 1471; HRMS (ES): calcd for $\text{C}_{20}\text{H}_{19}\text{NO}_2$ [M] $^+$: 305.1416; found: 305.1412.

Dihydrobenzofuran-Appended Oxindole **5a.** Colorless solid (8 mg, 20%); mp 134–136 °C; ^1H NMR (300 MHz, CDCl_3 , 25 °C) δ : 7.29 (d, $J = 7.6$ Hz, 1H), 7.17 (m, 2H), 6.91 (dd, $J = 6.9, 1.5$ Hz, 1H), 6.79 (t, $J = 7.6$ Hz, 1H), 6.74 (d, $J = 7.7$ Hz, 1H), 6.53 (d, $J = 8.3$

H₂, 1H), 5.53 and 4.99 (s, each 1H), 3.63 (s, 1H), 3.20 (s, 3H), 2.27 (s, 3H), 1.80 (s, 3H); ¹³C NMR (75 MHz, CDCl₃, 25 °C) δ: 174.0, 159.0, 150.8, 145.0, 131.5, 129.8, 128.2, 125.6, 125.2, 124.6, 121.7, 121.0, 109.8, 107.5, 101.0, 90.2, 52.8, 26.1, 24.8, 20.8; IR (CH₂Cl₂, cm⁻¹): ν 1709, 1612, 1487; HRMS (ES): calcd for C₂₀H₁₉NO₂ [M]⁺: 305.1416; found: 305.1419.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4b and 5b. From 43 mg (0.20 mmol) of α-allenol 1a, and after chromatography of the residue using hexanes/ethyl acetate (8:1) as eluent, 47 mg (67%) of the less polar compound 4b and 12 mg (17%) of the more polar compound 5b were obtained.

Dihydrobenzofuran-Appended Oxindole 4b. Pale yellow solid (47 mg, 67%); mp 90–92 °C; ¹H NMR (300 MHz, CDCl₃, 25 °C) δ: 7.19 (m, 4H), 6.82 (d, J = 8.5 Hz, 1H), 6.77 (m, 1H), 6.73 (t, J = 7.6 Hz, 1H), 5.12 and 4.76 (s, each 1H), 3.89 (s, 1H), 3.21 (s, 3H), 1.94 (s, 3H), 1.22 (s, 9H); ¹³C NMR (175 MHz, C₆D₆, 25 °C) δ: 173.0, 159.7, 149.7, 145.0, 143.8, 129.0, 128.9, 126.9, 126.7, 126.0, 122.5, 118.1, 110.1, 108.0, 100.3, 92.0, 54.5, 34.6, 31.9 (3C), 27.2, 25.9; IR (CH₂Cl₂, cm⁻¹): ν 1700, 1610, 1489; HRMS (ES): calcd for C₂₃H₂₅NO₂ [M]⁺: 347.1885; found: 347.1888.

Dihydrobenzofuran-Appended Oxindole 5b. Yellow oil (12 mg, 17%); ¹H NMR (700 MHz, CDCl₃, 25 °C) δ: 7.39 (d, J = 1.8 Hz, 1H), 7.31 (d, J = 7.4 Hz, 1H), 7.18 (t, J = 7.7 Hz, 1H), 7.16 (dd, J = 8.5, 2.0 Hz, 1H), 6.80 (t, J = 7.6 Hz, 1H), 6.74 (d, J = 7.7 Hz, 1H), 6.59 (d, J = 8.5 Hz, 1H), 5.58 and 5.05 (s, each 1H), 3.64 (s, 1H), 3.20 (s, 3H), 1.77 (s, 3H), 1.29 (s, 9H); ¹³C NMR (75 MHz, CDCl₃, 25 °C) δ: 174.0, 158.7, 150.9, 144.9, 143.6, 128.2, 128.1, 125.8, 124.8, 124.7, 121.7, 117.3, 109.5, 107.5, 101.1, 90.3, 52.7, 34.3, 31.6 (3C), 26.2, 24.5; IR (CH₂Cl₂, cm⁻¹): ν 1708, 1612, 1487; HRMS (ES): calcd for C₂₃H₂₅NO₂ [M]⁺: 347.1885; found: 347.1889.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4c and 5c. From 50 mg (0.23 mmol) of α-allenol 1a, and after chromatography of the residue using hexanes/ethyl acetate (5:1) as eluent, 43 mg (50%) of the less polar compound 4c and 9 mg (10%) of the more polar compound 5c (containing ca. 40% of its epimer 4c) were obtained.

Dihydrobenzofuran-Appended Oxindole 4c. Yellow solid (43 mg, 50%); mp 164–166 °C; ¹H NMR (300 MHz, C₆D₆, 25 °C) δ: 7.03 (dd, J = 8.5, 2.2 Hz, 1H), 6.92 (d, J = 2.2 Hz, 1H), 6.88 (s, 1H), 6.80 (t, J = 7.5 Hz, 1H), 6.56 (t, J = 7.6 Hz, 1H), 6.45 (d, J = 8.5 Hz, 1H), 6.13 (d, J = 7.7 Hz, 1H), 4.87 and 4.75 (s, each 1H), 3.58 (s, 1H), 2.68 (s, 3H), 2.0 (s, 3H); ¹³C NMR (175 MHz, C₆D₆, 25 °C) δ: 172.6, 160.3, 147.5, 145.0, 141.9, 133.9, 129.4, 126.8, 124.7, 123.1, 122.6, 113.4, 112.0, 108.1, 102.8, 92.7, 54.4, 26.7, 25.9; IR (CH₂Cl₂, cm⁻¹): ν 3334, 1682, 1489; HRMS (ES): calcd for C₁₉H₁₆BrNO₂ [M]⁺: 369.0364; found: 369.0370.

Dihydrobenzofuran-Appended Oxindole 5c. Yellow solid (9 mg, 10%; containing ca. 40% of its epimer 4c); mp 124–126 °C; ¹H NMR (300 MHz, CDCl₃, 25 °C) δ: 7.48 (d, J = 2.0 Hz, 0.4H, m), 7.20 (m, 3.6H, M + m), 6.77 (m, 2.6H, M + m), 6.52 (d, J = 8.5 Hz, 0.4H, m), 5.56 (s, 0.4H, m), 5.13 (s, 0.6H, M), 5.03 (s, 0.4H, m), 4.85 (s, 0.6H, M), 3.87 (s, 0.6H, M), 3.63 (s, 0.4H, m), 3.20 (s, 3H, M + m), 1.95 (s, 1.8H, M), 1.82 (s, 1.2H, m); ¹³C NMR (75 MHz, CDCl₃, 25 °C) δ: 173.6 (m), 172.7 (M), 159.3 (M + m), 149.3 (m), 146.6 (M), 144.9 (m), 144.1 (M), 133.3 (m), 133.2 (M), 128.4 (M), 128.3 (M + m), 126.2 (M + m), 125.5 (m), 124.6 (M), 124.1 (m), 123.7 (M + m), 122.3 (M), 121.9 (m), 112.7 (M), 112.6 (m), 111.9 (m), 111.5 (M), 107.8 (M), 107.7 (m), 102.9 (m), 102.1 (M), 91.9 (M), 91.1 (m), 53.9 (M), 52.7 (m), 26.2 (m), 26.0 (M), 24.9 (m); IR (CH₂Cl₂, cm⁻¹): ν 3299, 1706, 1466; HRMS (ES): calcd for C₁₉H₁₆BrNO₂ [M]⁺: 369.0364; found: 369.0357.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4d and 5d. From 52 mg (0.24 mmol) of α-allenol 1a, and after chromatography of the residue using hexanes/ethyl acetate (8:1) as the eluent, 33 mg (42%) of the less polar compound 4d and 15 mg (20%) of the more polar compound 5d (containing ca. 20% of its epimer 4d) were obtained.

Dihydrobenzofuran-Appended Oxindole 4d. Yellow oil; ¹H NMR (300 MHz, CDCl₃, 25 °C) δ: 7.18 (d, J = 7.5 Hz, 1H), 7.12 (d, J = 7.7 Hz, 1H), 6.75 (m, 4H), 6.61 (d, J = 1.6 Hz, 1H), 5.09 and 4.79

(s, each 1H), 3.88 (s, 1H), 3.70 (s, 3H), 3.20 (s, 3H), 1.94 (s, 3H), 1.94 (s, 3H); ¹³C NMR (175 MHz, CDCl₃, 25 °C) δ: 173.1, 154.8, 148.3, 144.1, 138.3, 128.1, 126.3, 125.0, 122.2, 117.6, 114.1, 110.2, 107.5, 104.9, 100.5, 91.2, 54.8, 54.9, 26.1, 26.0; IR (CH₂Cl₂, cm⁻¹): ν 1706, 1612, 1482; HRMS (ES): calcd for C₂₀H₁₉NO₃ [M]⁺: 321.1365; found: 321.1373.

Dihydrobenzofuran-Appended Oxindole 5d. Yellow oil; ¹H NMR (700 MHz, C₆D₆, 25 °C) δ: 7.30 (d, J = 7.5 Hz, 0.8H, m), 7.26 (d, J = 8.1 Hz, 0.2H, M), 7.04 (s, 0.2H, M), 6.93 (t, J = 7.7 Hz, 0.2H, M), 6.89 (m, 1.6H, m), 6.74 (d, J = 8.7 Hz, 0.2H, M), 6.70 (t, J = 7.2 Hz, 0.2H, M), 6.61 (t, J = 7.3 Hz, 0.8H, m), 6.48 (d, J = 2.6 Hz, 0.2H, M), 6.43 (d, J = 1.4 Hz, 1.6H, m), 6.20 (d, J = 7.7 Hz, 0.8H, m), 6.17 (d, J = 7.8 Hz, 0.2H, M), 5.31 (s, 0.8H, m), 4.99 and 4.97 (s, each 0.2H, M), 4.79 (s, 0.8H, m), 3.33 (s, 0.8H, m), 3.23 (s, 3H, M + m), 3.11 (s, 0.2H, M), 2.71 (s, 2.4H, m), 2.65 (s, 0.6H, M), 1.97 (s, 2.4H, m), 1.84 (s, 0.6H, M); ¹³C NMR (175 MHz, C₆D₆, 25 °C) δ: 173.9 (M+m), 156.7 (m), 155.1 (M), 152.8 (m), 150.4 (M), 146.0 (m), 143.6 (M), 129.7 (m), 128.0 (m), 127.5 (M), 127.0 (M), 126.6 (M), 126.1 (m), 125.6 (M), 123.1 (m), 122.5 (M), 122.4 (m), 121.9 (m), 121.3 (M), 118.7 (M), 118.2 (m), 111.5 (m), 111.0 (M), 108.3 (M), 108.0 (m), 105.9 (m), 105.6 (M), 101.4 (m), 100.8 (M), 91.5 (M + m), 55.6 (m), 55.4 (M), 53.2 (M + m), 26.1 (M), 26.0 (m), 25.6 (m), 25.5 (M); IR (CH₂Cl₂, cm⁻¹): ν 1708, 1611, 1482; HRMS (ES): calcd for C₂₀H₁₉NO₃ [M]⁺: 321.1365; found: 321.1372.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4e and 5e. From 40 mg (0.20 mmol) of α-allenol 1b, and after chromatography of the residue using dichloromethane/ethyl acetate (40:1) as eluent, 35 mg (60%) of the less polar compound 4e and 14 mg (24%) of the more polar compound 5e were obtained.

Dihydrobenzofuran-Appended Oxindole 4e. Colorless solid (35 mg, 60%); mp 166–168 °C; ¹H NMR (300 MHz, CDCl₃, 25 °C) δ: 7.88 (br s, 1H), 7.17 (d, J = 7.5 Hz, 1H), 7.07 (t, J = 7.8 Hz, 1H), 7.00 (d, J = 8.2 Hz, 1H), 6.95 (s, 1H), 6.75 (m, 3H), 5.18 and 4.92 (s, each 1H), 3.90 (s, 1H), 2.21 (s, 3H), 1.92 (s, 3H); ¹³C NMR (75 MHz, CDCl₃, 25 °C) δ: 174.9, 158.4, 147.9, 141.0, 131.4, 129.8, 128.1, 126.7, 126.0, 125.7, 122.2, 121.1, 109.5, 109.0, 100.4, 90.8, 54.4, 26.1, 20.7; IR (CH₂Cl₂, cm⁻¹): ν 3252, 1705, 1482; HRMS (ES): calcd for C₁₉H₁₇NO₂ [M]⁺: 291.1259; found: 291.1246.

Dihydrobenzofuran-Appended Oxindole 5e. Colorless solid (14 mg, 24%); mp 164–166 °C; ¹H NMR (300 MHz, CDCl₃, 25 °C) δ: 7.53 (br s, 1H), 7.28 (m, 1H), 7.19 (s, 1H), 7.11 (t, J = 7.7 Hz, 1H), 6.92 (dd, J = 7.7, 1.4 Hz, 1H), 6.78 (m, 2H), 6.57 (d, J = 8.2 Hz, 1H), 5.53 and 4.97 (s, each 1H), 3.65 (s, 1H), 2.27 (s, 3H), 1.80 (s, 3H); ¹³C NMR (75 MHz, CDCl₃, 25 °C) δ: 175.6, 159.0, 150.5, 141.8, 131.6, 129.9, 128.3, 126.1, 125.8, 125.2, 121.8, 121.1, 109.8, 109.0, 101.2, 90.1, 53.1, 24.7, 20.8; IR (CH₂Cl₂, cm⁻¹): ν 3253, 1708, 1480; HRMS (ES): calcd for C₁₉H₁₇NO₂ [M]⁺: 291.1259; found: 291.1250.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4f and 5f. From 40 mg (0.20 mmol) of α-allenol 1b, and after chromatography of the residue using hexanes/ethyl acetate (6:1) as eluent, 35 mg (52%) of the less polar compound 4f and 4 mg (6%) of the more polar compound 5f (containing ca. 50% of its epimer 4f) were obtained.

Dihydrobenzofuran-Appended Oxindole 4f. Pale yellow solid (35 mg, 52%); mp 203–205 °C; ¹H NMR (700 MHz, C₆D₆, 25 °C) δ: 7.30 (d, J = 7.5 Hz, 1H), 7.27 (s, 1H), 7.09 (dd, J = 8.5, 2.0 Hz, 1H), 7.05 (s, 1H), 6.85 (d, J = 8.5 Hz, 1H), 6.70 (t, J = 7.7 Hz, 1H), 6.49 (t, J = 7.2 Hz, 1H), 6.04 (m, 1H), 5.17 and 5.13 (s, each 1H), 3.66 (s, 1H), 2.06 (s, 3H), 1.07 (s, 9H); ¹³C NMR (175 MHz, CDCl₃, 25 °C) δ: 174.2, 159.6, 149.5, 143.8 (2C), 142.0, 129.0, 128.0, 127.3, 126.7, 126.6, 122.4, 118.2, 110.1, 109.3, 100.6, 91.9, 54.7, 34.6, 31.9 (3C), 27.0; IR (CH₂Cl₂, cm⁻¹): ν 3214, 1706, 1487; HRMS (ES): calcd for C₂₂H₂₃NO₂ [M]⁺: 333.1729; found: 333.1721.

Dihydrobenzofuran-Appended Oxindole 5f. Yellow solid (4 mg, 6%; containing ca. 50% of its epimer 4f); mp 182–184 °C; ¹H NMR (300 MHz, C₆D₆, 25 °C) δ: 8.20 (br s, 0.5H, m), 8.14 (br s, 0.5H, M), 7.45 (d, J = 1.9 Hz, 0.5H, m), 7.30 (m, 1.5H, 2M+m), 7.16 (m, 0.5H, M), 7.10 (m, 1H, 2M), 6.81 (m, 2H, 2M + 2m), 6.54 (m, 1H, M + m), 6.35 (d, J = 7.7 Hz, 0.5H, m), 6.27 (d, J = 7.7 Hz, 0.5H, m), 5.41 (s, 0.5H, m), 5.18 and 5.15 (s, each 0.5H, M), 4.79 (s, 0.5H,

m), 3.70 (s, 0.5H, M), 3.35 (s, 0.5H, m), 2.06 (s, 1.5H, M), 1.90 (s, 1.5H, m), 1.15 (s, 4.5H, m), 1.08 (s, 4.5H, M); ^{13}C NMR (75 MHz, CDCl_3 , 25 °C) δ : 176.6 (m), 175.7 (M), 160.2 (m), 159.6 (M), 152.3 (m), 149.5 (M), 144.0 (m), 143.9 (M), 143.3 (m), 142.3 (M), 129.1 (m), 129.0 (M), 128.7 (m), 128.0 (M), 127.2 (M), 126.8 (m), 126.7 (M), 126.6 (M), 126.3 (m), 125.7 (m), 122.5 (M), 121.9 (m), 118.2 (M), 117.8 (m), 110.7 (m), 110.1 (M), 109.8 (m), 109.7 (M), 100.9 (m), 100.7 (M), 91.9 (M), 91.3 (m), 55.0 (M), 53.6 (m), 34.7 (m), 34.6 (M), 32.0 (3C, m), 31.9 (3C, M), 27.1 (M), 25.6 (m); IR (CH_2Cl_2 , cm^{-1}): ν 3212, 1704, 1486; HRMS (ES): calcd for $\text{C}_{22}\text{H}_{23}\text{NO}_2$ [M] $^+$: 333.1729; found: 333.1710.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4g and 5g. From 40 mg (0.20 mmol) of α -allenol 1b, and after chromatography of the residue using hexanes/ethyl acetate (4:1) as eluent, 27 mg (38%) of the less polar compound 4g and 4 mg (6%) of the more polar compound 5g (containing ca. 20% of its epimer 4g) were obtained.

Dihydrobenzofuran-Appended Oxindole 4g. Pale yellow solid (27 mg, 38%); mp 204–206 °C; ^1H NMR (700 MHz, C_6D_6 , 25 °C) δ : 7.10 (dd, $J = 8.2, 6.2$ Hz, 1H), 7.02 (dd, $J = 8.5, 2.0$ Hz, 1H), 6.95 (d, $J = 2.0$ Hz, 1H), 6.72 (t, $J = 7.7$ Hz, 1H), 6.50 (t, $J = 7.7$ Hz, 1H), 6.44 (d, $J = 8.5$ Hz, 1H), 6.32 (d, $J = 8.7$ Hz, 1H), 6.12 (d, $J = 7.7$ Hz, 1H), 4.99 and 4.83 (s, each 1H), 3.54 (s, 1H), 1.93 (s, 3H); ^{13}C NMR (175 MHz, C_6D_6 , 25 °C) δ : 174.0, 160.3, 147.4, 142.1, 133.9, 133.0, 129.4, 127.1, 126.2, 124.8, 122.4, 113.5, 112.0, 109.5, 103.1, 92.5, 54.7, 26.6; IR (CH_2Cl_2 , cm^{-1}): ν 3256, 1699, 1464; HRMS (ES): calcd for $\text{C}_{18}\text{H}_{14}\text{BrNO}_2$ [M] $^+$: 355.0208; found: 355.0192.

Dihydrobenzofuran-Appended Oxindole 5g. Yellow solid (4 mg, 6%; containing ca. 20% of its epimer 4g); mp 194–196 °C; ^1H NMR (300 MHz, C_6D_6 , 25 °C) δ : 7.46 (m, 0.2H, m), 7.31 (d, $J = 2.0$ Hz, 0.8H, M), 7.11 (m, 1H, M + m), 7.02 (dd, $J = 8.5, 2.0$ Hz, 1H, M + m), 6.96 (d, $J = 2.0$ Hz, 1H, M + m), 6.77 (m, 1H, M + m), 6.52 (td, $J = 7.7, 1.0$ Hz, 1H, M + m), 6.44 (d, $J = 8.5$ Hz, 1H, M + m), 6.20 (m, 1H, M + m), 5.09 (s, 0.2H, m), 5.00 and 4.84 (s, each 0.8H, M), 4.62 (s, 0.2H, m), 3.55 (s, 0.8H, M), 3.18 (s, 0.2H, m), 1.93 (s, 2.4H, M), 1.79 (s, 0.6H, m); ^{13}C NMR (75 MHz, C_6D_6 , 25 °C) δ : 174.6 (M + m), 160.3 (M + m), 147.4 (M + m), 142.2 (M + m), 134.1 (m), 133.9 (3C, 2M + 1m), 129.4 (M), 128.9 (m), 127.1 (M), 126.4 (m), 126.2 (M), 125.7 (m), 124.8 (M + m), 124.5 (m), 122.5 (M), 122.0 (m), 113.5 (M), 112.7 (m), 112.0 (M), 109.7 (M + m), 103.1 (M + m), 92.5 (M), 91.9 (m), 54.8 (M), 53.4 (m), 26.6 (M), 25.4 (m); IR (CH_2Cl_2 , cm^{-1}): ν 3187, 1708, 1464; HRMS (ES): calcd for $\text{C}_{18}\text{H}_{14}\text{BrNO}_2$ [M] $^+$: 355.0208; found: 355.0218.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4h and 5h. From 43 mg (0.21 mmol) of α -allenol 1b, and after chromatography of the residue using hexanes/ethyl acetate (7:1) as eluent, 32 mg (55%) of the less polar compound 4h and 6 mg (10%) of the more polar compound 5h (containing ca. 40% of its epimer 4h) were obtained.

Dihydrobenzofuran-Appended Oxindole 4h. Colorless solid (32 mg, 55%); mp 164–166 °C; ^1H NMR (700 MHz, CDCl_3 , 25 °C) δ : 7.65 (s, 1H), 7.20 (td, $J = 8.4, 1.3$ Hz, 1H), 7.16 (d, $J = 7.5$ Hz, 1H), 7.14 (dd, $J = 7.6, 0.8$ Hz, 1H), 7.06 (t, $J = 7.7$ Hz, 1H), 6.90 (d, $J = 8.2$ Hz, 1H), 6.78 (td, $J = 7.5, 0.8$ Hz, 1H), 6.73 (d, $J = 7.7$ Hz, 1H), 6.70 (td, $J = 7.7, 0.8$ Hz, 1H), 5.22 and 4.96 (s, each 1H), 3.92 (s, 1H), 1.94 (s, 3H); ^{13}C NMR (175 MHz, CDCl_3 , 25 °C) δ : 174.6, 160.3, 147.6, 140.9, 130.6, 128.2, 126.7, 126.2, 125.6, 122.2, 120.8, 120.6, 109.9, 108.9, 100.9, 90.7, 50.4, 26.0; IR (CH_2Cl_2 , cm^{-1}): ν 3244, 1702, 1468; HRMS (ES): calcd for $\text{C}_{18}\text{H}_{15}\text{NO}_2$ [M] $^+$: 277.1103; found: 277.1105.

Dihydrobenzofuran-Appended Oxindole 5h. Yellow solid (6 mg, 10%; containing ca. 40% of its epimer 4h); mp 124–126 °C; ^1H NMR (300 MHz, CDCl_3 , 25 °C) δ : 7.91 (m, 1H, M + m), 7.38 (dd, $J = 7.6, 0.9$ Hz, 0.4H, m), 7.29–7.04 (m, 3.6H, M + m), 6.90 (d, $J = 8.0$ Hz, 0.6H, M), 6.86 (dd, $J = 7.5, 0.8$ Hz, 0.4H, m), 6.82–6.65 (m, 3H, M + m), 5.58 (s, 0.4H, m), 5.22 (s, 0.6H, M), 5.02 (s, 0.4H, m), 4.95 (s, 0.6H, M), 3.92 (s, 0.6H, M), 3.66 (s, 0.4H, m), 1.94 (s, 1.8H, M), 1.83 (s, 1.2H, m); ^{13}C NMR (75 MHz, CDCl_3 , 25 °C) δ : 174.5 (M + m), 160.3 (M + m), 147.5 (M + m), 140.9 (M + m), 130.6 (M + m), 128.4 (m), 128.2 (M), 126.8 (m), 126.7 (M), 126.2 (M + m), 125.5 (M + m), 122.3 (m), 122.1 (M), 120.8 (M + m), 120.6 (M + m),

109.9 (M + m), 108.9 (M + m), 100.9 (M + m), 90.7 (M + m), 54.3 (M + m), 26.0 (M + m); IR (CH_2Cl_2 , cm^{-1}): ν 3231, 1707, 1468; HRMS (ES): calcd for $\text{C}_{18}\text{H}_{15}\text{NO}_2$ [M] $^+$: 277.1103; found: 277.1111.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4i and 5i. From 50 mg (0.18 mmol) of α -allenol 1c, and after chromatography of the residue using hexanes/ethyl acetate (4:1) as eluent, 33 mg (49%) of the less polar compound 4i and 6 mg (9%) of the more polar compound 5i (containing ca. 50% of its epimer 4i) were obtained.

Dihydrobenzofuran-Appended Oxindole 4i. Yellow oil (33 mg, 49%); ^1H NMR (300 MHz, C_6D_6 , 25 °C) δ : 7.42 (s, 1H), 6.89–6.66 (m, 4H), 5.72 (d, $J = 8.2$ Hz, 1H), 5.08 and 5.03 (s, each 1H), 3.47 (s, 1H), 1.97 (s, 3H), 1.88 (s, 3H); ^{13}C NMR (175 MHz, C_6D_6 , 25 °C) δ : 173.6, 159.7, 149.1, 141.1, 132.6 (2C), 131.5, 131.0, 130.6, 126.9, 122.3, 115.1, 110.7, 110.3, 101.0, 91.4, 55.1, 26.7, 21.0; IR (CH_2Cl_2 , cm^{-1}): ν 3249, 1707, 1485; HRMS (ES): calcd for $\text{C}_{19}\text{H}_{16}\text{BrNO}_2$ [M] $^+$: 369.0364; found: 369.0347.

Dihydrobenzofuran-Appended Oxindole 5i. Yellow oil (6 mg, 9%; containing ca. 50% of its epimer 4i); ^1H NMR (300 MHz, C_6D_6 , 25 °C) δ : 7.42 (s, 1H, M + m), 7.22–6.66 (m, 4H, M + m), 6.58 (d, $J = 7.4$ Hz, 0.5H, m), 6.46 (d, $J = 8.2$ Hz, 0.5H, m), 5.73 (d, $J = 8.3$ Hz, 0.5H, M), 5.69 (d, $J = 8.3$ Hz, 0.5H, m), 5.26 (s, 0.5H, m), 5.07 and 5.03 (s, each 0.5H, M), 4.69 (s, 0.5H, m), 3.47 (s, 0.5H, M), 3.08 (s, 0.5H, m), 2.00 (s, 1.5H, m), 1.97 (s, 1.5H, M), 1.88 (s, 1.5H, M), 1.80 (s, 1.5H, m); ^{13}C NMR (75 MHz, C_6D_6 , 25 °C) δ : 174.4 (M + m), 159.5 (M + m), 149.0 (M + m), 141.1 (M + m), 132.5, (4C, 2M + 2m), 131.4 (M + m), 130.8 (M + m), 130.5 (M), 129.2 (m), 127.5 (m), 126.7 (M), 122.1 (M + m), 115.5 (m), 115.1 (M), 110.8 (M + m), 110.1 (M + m), 100.9 (M + m), 91.2 (M + m), 60.4 (m), 55.2 (M), 26.6 (M + m), 21.0 (M + m); IR (CH_2Cl_2 , cm^{-1}): ν 3245, 1703, 1475; HRMS (ES): calcd for $\text{C}_{19}\text{H}_{16}\text{BrNO}_2$ [M] $^+$: 369.0364; found: 369.0383.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4j and 5j. From 40 mg (0.17 mmol) of α -allenol 1d, and after chromatography of the residue using hexanes/ethyl acetate (6:1) as eluent, 38 mg (69%) of the less polar compound 4j and 4 mg (7%) of the more polar compound 5j were obtained.

Dihydrobenzofuran-Appended Oxindole 4j. Colorless solid (38 mg, 69%); mp 103–105 °C; ^1H NMR (700 MHz, CDCl_3 , 25 °C) δ : 7.76 (br s, 1H), 7.00 (d, $J = 8.1$ Hz, 1H), 6.94 (s, 1H), 6.68 (m, 2H), 6.67 (m, 2H), 5.17 and 4.93 (s, each 1H), 3.93 (s, 1H), 3.80 (s, 3H), 2.21 (s, 3H), 1.92 (s, 3H); ^{13}C NMR (175 MHz, CDCl_3 , 25 °C) δ : 174.2, 158.4, 147.8, 143.0, 131.3, 129.9, 129.7, 126.5, 126.0, 122.5, 121.0, 119.0, 110.3, 109.5, 100.5, 90.8, 55.4, 55.1, 29.6, 26.0; IR (CH_2Cl_2 , cm^{-1}): ν 3228, 1709, 1493, 1462; HRMS (ES): calcd for $\text{C}_{20}\text{H}_{19}\text{NO}_3$ [M] $^+$: 321.1365; found: 321.1358.

Dihydrobenzofuran-Appended Oxindole 5j. Colorless oil (4 mg, 7%); ^1H NMR (700 MHz, CDCl_3 , 25 °C) δ : 7.43 (s, 1H), 7.19 (s, 1H), 6.93 (m, 1H), 6.76 (m, 1H), 6.71 (d, $J = 8.2$ Hz, 1H), 6.59 (d, $J = 8.2$ Hz, 1H), 5.52 and 4.96 (s, each 1H), 3.82 (s, 3H), 3.67 (s, 1H), 2.27 (s, 3H), 1.79 (s, 3H); ^{13}C NMR (175 MHz, CDCl_3 , 25 °C) δ : 174.8, 158.9, 150.4, 143.0, 131.5, 130.8, 129.9, 125.9, 125.3, 122.1, 121.1, 118.5, 110.5, 109.9, 101.2, 90.1, 55.5, 53.7, 29.7, 24.7; IR (CH_2Cl_2 , cm^{-1}): ν 3221, 1704, 1494, 1461; HRMS (ES): calcd for $\text{C}_{20}\text{H}_{19}\text{NO}_3$ [M] $^+$: 321.1365; found: 321.1357.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4k and 5k. From 40 mg (0.17 mmol) of α -allenol 1d, and after chromatography of the residue using hexanes/ethyl acetate (4:1) as eluent, 29 mg (48%) of the less polar compound 4k and 6 mg (9%) of the more polar compound 5k (containing ca. 50% of its epimer 4k) were obtained.

Dihydrobenzofuran-Appended Oxindole 4k. Pale yellow solid (29 mg, 48%); mp 193–195 °C; ^1H NMR (300 MHz, C_6D_6 , 25 °C) δ : 8.44 (m, 1H), 7.13 (m, 2H), 7.03 (d, $J = 7.5$ Hz, 1H), 6.89 (dd, $J = 8.3, 0.7$ Hz, 1H), 6.47 (t, $J = 8.0$ Hz, 1H), 6.10 (d, $J = 8.3$ Hz, 1H), 5.23 and 5.16 (s, each 1H), 3.80 (s, 1H), 3.05 (s, 3H), 2.12 (s, 3H), 1.09 (s, 9H); ^{13}C NMR (75 MHz, CDCl_3 , 25 °C) δ : 174.9, 159.7, 149.4, 143.9, 143.8, 131.4, 129.0, 127.5, 126.7, 122.9, 119.7, 118.2, 111.1, 110.1, 100.9, 92.0, 55.7, 55.1, 34.6, 31.9 (3C), 27.0; IR (CH_2Cl_2 , cm^{-1}): ν

3225, 1708, 1490; HRMS (ES): calcd for $C_{23}H_{25}NO_3$ $[M]^+$: 363.1834; found: 363.1841.

Dihydrobenzofuran-Appended Oxindole 5k. Yellow solid (6 mg, 9%; containing ca. 50% of its epimer **4k**); mp 181–183 °C; 1H NMR (300 MHz, C_6D_6 , 25 °C) δ : 8.22 (br s, 0.5H, m), 8.12 (br s, 0.5H, M), 7.47 (d, $J = 2.0$ Hz, 0.5H, m), 7.13 (m, 1.5H, M + m + M), 7.01 (m, 1H, M + m), 6.91 (m, 1H, M + m), 6.55 (m, 1H, m), 6.48 (t, $J = 8.0$ Hz, 0.5H, M), 6.22 (d, $J = 8.3$ Hz, 0.5H, m), 6.10 (d, $J = 8.3$ Hz, 0.5H, M), 5.45 (s, 0.5H, m), 5.21 and 5.15 (s, each 0.5H, M), 4.868 (s, 0.5H, m), 3.79 (s, 0.5H, M), 3.42 (s, 0.5H, m), 3.12 (s, 1.5H, m), 3.04 (s, 1.5H, M), 2.12 (s, 1.5H, M), 1.94 (s, 1.5H, m), 1.16 (s, 4.5H, m), 1.09 (s, 4.5H, M); ^{13}C NMR (75 MHz, C_6D_6 , 25 °C) δ : 175.6 (m), 174.8 (M), 160.2 (M), 159.7 (M), 152.5 (m), 149.4 (M), 143.9 (m), 143.9 (m), 143.7 (2C, M), 132.3 (m), 131.3 (M), 129.1 (m), 129.0 (M), 127.5 (M), 127.0 (m), 126.7 (M), 125.8 (m), 122.9 (M), 122.2 (m), 119.7 (M), 119.3 (m), 118.2 (M), 117.8 (m), 111.1 (M), 111.0 (m), 110.8 (m), 110.1 (M), 101.0 (m), 100.8 (M), 92.0 (M), 91.4 (m), 55.7 (M), 55.1 (m), 55.1 (M), 54.1 (m), 34.7 (M), 32.0 (3C, m), 31.9 (3C, M), 30.6 (m), 27.1 (M), 25.5 (m); IR (CH_2Cl_2 , cm^{-1}): ν 3220, 1710, 1391; HRMS (ES): calcd for $C_{23}H_{25}NO_3$ $[M]^+$: 363.1834; found: 363.1845.

Preparation of Dihydrobenzofuran-Appended Oxindoles 4l and 5l. From 35 mg (0.17 mmol) of α -allenol **1b**, and after chromatography of the residue using hexanes/ethyl acetate (6:1) as eluent, 26 mg (50%) of the less polar compound **4l** and 9 mg (17%) of the more polar compound **5l** (containing ca. 40% of its epimer **4l**) were obtained.

Dihydrobenzofuran-Appended Oxindole 4l. Colorless oil; 1H NMR (700 MHz, C_6D_6 , 25 °C) δ : 7.31 (d, $J = 7.5$ Hz, 1H), 6.77 (t, $J = 7.5$ Hz, 1H), 6.73 (d, $J = 8.7$ Hz, 1H), 6.70 (dd, $J = 8.7, 2.6$ Hz, 1H), 6.57 (t, $J = 7.6$ Hz, 1H), 6.51 (m, 1H), 6.29 (m, 1H), 5.13 and 5.06 (s, each 1H), 3.69 (m, 1H), 3.12 (s, 3H), 2.1 (s, 3H); ^{13}C NMR (175 MHz, C_6D_6 , 25 °C) δ : 175.8, 155.9, 155.0, 149.5, 142.3, 128.7, 127.5, 127.4, 126.7, 122.5, 118.7, 111.0, 109.7, 105.6, 101.2, 91.8, 55.4, 55.1, 26.9; IR (CH_2Cl_2 , cm^{-1}): ν 3263, 1706, 1482; HRMS (ES): calcd for $C_{19}H_{17}NO_3$ $[M]^+$: 307.1208; found: 307.1218.

Dihydrobenzofuran-Appended Oxindole 5l. Yellow oil; 1H NMR (700 MHz, C_6D_6 , 25 °C) δ : 7.31 (d, $J = 7.5$ Hz, 0.4H, M), 7.24 (d, $J = 7.5$ Hz, 0.6H, m), 7.23 (d, $J = 7.7$ Hz, 0.6H, m), 6.87 (m, 0.6H, m), 6.77 (t, $J = 7.7$ Hz, 0.4H, M), 6.73 (d, $J = 8.7$ Hz, 0.4H, M), 6.70 (dd, $J = 8.7, 2.6$ Hz, 0.4H, M), 6.66 (t, $J = 7.7$ Hz, 0.6H, m), 6.58 (m, 1H, M + m), 6.51 (m, 0.4H, M), 6.45 (m, 0.6H, m), 6.35 (m, 2H, M + m), 5.30 (s, 0.6H, m), 5.12 and 5.05 (s, each 0.4H, M), 4.75 (s, 0.6H, m), 3.69 (s, 0.4H, M), 3.32 (s, 0.6H, m), 3.23 (s, 1.8H, m), 3.13 (s, 1.2H, M), 2.37 (s, 1.8H, m), 2.07 (s, 1.2H, m); ^{13}C NMR (175 MHz, C_6D_6 , 25 °C) δ : 176.8 (m), 175.8 (M), 156.6 (m), 155.9 (M), 155.2 (M), 152.4 (m), 151.1 (m), 149.5 (M), 143.4 (m), 142.5 (M), 141.2 (m), 129.8 (m), 128.7 (M), 127.5 (M), 127.3 (M), 126.5 (M), 126.3 (m), 123.5 (m), 122.5 (M), 122.0 (m), 118.7 (M), 118.2 (m), 111.5 (m), 111.0 (M), 110.4 (m), 109.8 (M), 105.9 (m), 105.6 (M), 101.5 (m), 101.3 (M), 91.8 (M), 91.3 (m), 55.6 (m), 55.4 (M), 55.2 (M), 53.7 (m), 28.0 (m), 27.0 (M); IR (CH_2Cl_2 , cm^{-1}): ν 3263, 1707, 1480; HRMS (ES): calcd for $C_{19}H_{17}NO_3$ $[M]^+$: 307.1208; found: 307.1210.

Ketone 7a. From 44 mg (0.16 mmol) of the allenol **6a** and after chromatography of the residue using hexanes/ethyl acetate (5:1) as eluent, compound **7a** (14 mg, 30%) was obtained as a yellow oil; 1H NMR (300 MHz, $CDCl_3$, 25 °C) δ : 7.27 (m, 1H), 7.00 (m, 2H), 6.79 (d, $J = 8.1$ Hz, 1H), 6.53 (m, 3H), 3.00 (s, 3H), 2.24 (s, 3H); ^{13}C NMR (75 MHz, $CDCl_3$, 25 °C) δ : 204.0, 166.7, 151.4, 144.5, 132.8, 130.0, 129.9, 129.8, 129.2, 127.8, 123.8, 123.1, 121.9, 120.6, 115.0, 108.2, 29.0, 25.9; IR (CH_2Cl_2 , cm^{-1}): ν 2926, 1697, 1606, 1487; HRMS (ES): calcd for $C_{18}H_{15}NO_2$ $[M]^+$: 277.1103; found: 277.1104.

ASSOCIATED CONTENT

Supporting Information

ORTEP drawing of compound **5a** as well as copies of the 1H NMR and ^{13}C NMR spectra for all new compounds. This

material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) For selected reviews, see: (a) Donnelly, D. M. X.; Meegan, M. J. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R., Rees, C. W., Eds.; Pergamon: Oxford, U.K., 1984; Vol. 4. (b) Ayers, A. C.; Loike, J. K. *Lignans: Chemical, Biological and Clinical Properties*; Cambridge, 1990. (c) Watzke, A.; O'Malley, S. J.; Bergman, R. G.; Ellman, J. A. *J. Nat. Prod.* **2006**, *69*, 1231. (d) Veitch, N. C. *Nat. Prod. Rep.* **2007**, *24*, 417. (e) Shen, T.; Wang, X.-N.; Lou, H.-X. *Nat. Prod. Rep.* **2009**, *26*, 916.
- (2) For a review, see: Sheppard, T. D. *J. Chem. Res.* **2011**, *35*, 337.
- (3) For recent selected reviews, see: (a) Fensterbank, L.; Malacria, M. *Acc. Chem. Res.* **2014**, *47*, 953. (b) Alcaide, B.; Almendros, P. *Acc. Chem. Res.* **2014**, *47*, 939. (c) Lechel, T.; Pfengle, F.; Reissig, H.-U.; Zimmer, R. *ChemCatChem* **2013**, *5*, 2100. (d) Yu, S.; Ma, S. *Angew. Chem., Int. Ed.* **2012**, *51*, 3074. (e) Yang, W.; Hashmi, A. S. K. *Chem. Soc. Rev.* **2014**, *43*, 2941.
- (4) For the coupling of an allene intermediate with halogenated phenols, see: (a) Bi, H.-P.; Liu, X.-Y.; Gou, F.-R.; Guo, L.-N.; Duan, X.-H.; Shu, X.-Z.; Liang, Y.-M. *Angew. Chem., Int. Ed.* **2007**, *46*, 7068. For the condensation of allenolates and phenols, see: (b) Kim, S.; Kang, D.; Lee, C.-H.; Lee, P. H. *J. Org. Chem.* **2012**, *77*, 6530. For the gold-catalyzed chromane formation from allenamides and phenols, see: (c) Slater, N. H.; Brown, N. J.; Elsegood, M. R. J.; Kimber, M. C. *Org. Lett.* **2014**, *16*, 4606.
- (5) (a) Alcaide, B.; Almendros, P.; Martínez del Campo, T.; Quirós, M. T. *Chem.—Eur. J.* **2009**, *15*, 3344. (b) Alcaide, B.; Almendros, P.; Luna, A.; Cembellín, S.; Arnó, M.; Domingo, L. R. *Chem.—Eur. J.* **2011**, *17*, 11559. (c) Alcaide, B.; Almendros, P.; Luna, A.; Prieto, N. J. *Org. Chem.* **2012**, *77*, 11388.
- (6) Alcaide, B.; Almendros, P.; Quirós, M. T.; López, R.; Menéndez, M. I.; Sochacka-Ćwikła, A. *J. Am. Chem. Soc.* **2013**, *135*, 898.
- (7) (a) Hoffmann-Röder, A.; Krause, N. *Org. Lett.* **2001**, *3*, 2537. (b) Hashmi, A. S. K.; Blanco, M. C.; Fischer, D.; Bats, J. W. *Eur. J. Org. Chem.* **2006**, 1387. (c) Asikainen, M.; Krause, N. *Adv. Synth. Catal.* **2009**, *351*, 2305. (d) Brasholz, M.; Dugović, B.; Reissig, H.-U. *Synthesis* **2010**, 3855. (e) Alcaide, B.; Almendros, P.; Martínez del Campo, T.; Fernández, I. *Chem. Commun.* **2011**, *47*, 9054.
- (8) For a different reactivity of 3-hydroxy-2-oxindoles and phenols under Lewis acid catalysis, see: (a) Kinthada, L. K.; Ghosh, S.; Babu, K. N.; Sharique, M.; Biswas, S.; Bisai, A. *Org. Biomol. Chem.* **2014**, *12*, 8152. (b) Ghosh, S.; Kinthada, L. K.; Bhunia, S.; Bisai, A. *Chem. Commun.* **2012**, *48*, 10132. (c) Zhou, F.; Cao, Z.-Y.; Zhang, J.; Yang, H.-B.; Zhou, J. *Chem.—Asian J.* **2012**, *7*, 233. For a different reactivity of 3-hydroxy-2-oxindoles and phenols under organocatalysis, see: (d) Liu, Y.; Zhang, H.-H.; Zhang, Y.-C.; Jiang, Y.; Shi, F.; Tu, S.-J. *Chem. Commun.* **2014**, *50*, 12054. For a different reactivity of allenes and phenols, see: (e) Nemoto, T.; Nozaki, T.; Yoshida, M.; Hamada, Y. *Adv. Synth. Catal.* **2013**, *355*, 2693.

(9) Slightly improved yields were obtained by the use of Ga(OTf)₃ in comparison with Bi(OTf)₃, In(OTf)₃, Zn(OTf)₂, Fe(OTf)₃, and Yb(OTf)₃.

(10) For a review on the use of Ga(OTf)₃ in organic reactions, see: (a) Prakash, G. K. S.; Mathew, T.; Olah, G. A. *Acc. Chem. Res.* **2012**, *45*, 565. For a recent use, see: (b) Zhang, S.; Xu, Z.; Jia, J.; Tung, C.-H.; Xu, Z. *Chem. Commun.* **2014**, *50*, 12084.

(11) Compounds **4a–i** and **5a–i** can be considered as hybrid scaffolds as a combination of the biologically relevant oxindole and dihydrobenzofuran frameworks. For reviews on hybrid chemical entities, see: (a) Decker, M. *Curr. Med. Chem.* **2011**, *18*, 1464. (b) Tsogoeva, S. B. *Mini-Rev. Med. Chem.* **2010**, *10*, 773. (c) Meunier, B. *Acc. Chem. Res.* **2008**, *41*, 69. (d) Tietze, L. F.; Bell, H. P.; Chandrasekhar, S. *Angew. Chem., Int. Ed.* **2003**, *42*, 3996.

(12) CCDC-907155 contains the supplementary crystallographic data for this paper (www.ccdc.cam.ac.uk/data_request/cif).

(13) The transformation of isoindolinone-tethered alkoyallenols into oxopropylidene isoindolinones has been reported by treatment with aqueous sulphuric acid: (a) Kaden, S.; Reissig, H.-U.; Brüdgam, I.; Hartl, H. *Synthesis* **2006**, 1351. For the acid-catalyzed synthesis of α,β -disubstituted conjugated enones by a Meyer–Schuster-type rearrangement in allenols, see: (b) Alcaide, B.; Almendros, P.; Cembellin, S.; Martínez del Campo, T. *Adv. Synth. Catal.* **2015**, *357*, 1070.